



# Lab-Scale Development of a Solid Sorbent for CO<sub>2</sub> Capture Process for Coal-Fired Power Plants

DE-FE0026432

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# Lab-Scale Development of a Solid Sorbent for CO<sub>2</sub> Capture Process for Coal-Fired Power Plants

## **Project Details – DE-FE0026432**

**Funding:** \$1,989,415

- \$1,591,532 DOE
- \$ 397,883 Cost Share

**Period:** October 2015 – March 2018

## **Goals/Objective:**

- Develop novel 3<sup>rd</sup> generation fluidizable solid sorbents for RTI's sorbent-based CO<sub>2</sub> capture process:
  - ❖ **Fluidizable, hybrid-metal organic frameworks**
  - ❖ **Fluidizable hybrid-phosphorus dendrimers**

# Project Outline

## BP1

- **Design and synthesize two novel fluidizable CO<sub>2</sub> adsorbents.**
- **Demonstrate the superior performance of these advanced CO<sub>2</sub> solid sorbents at the lab scale.**

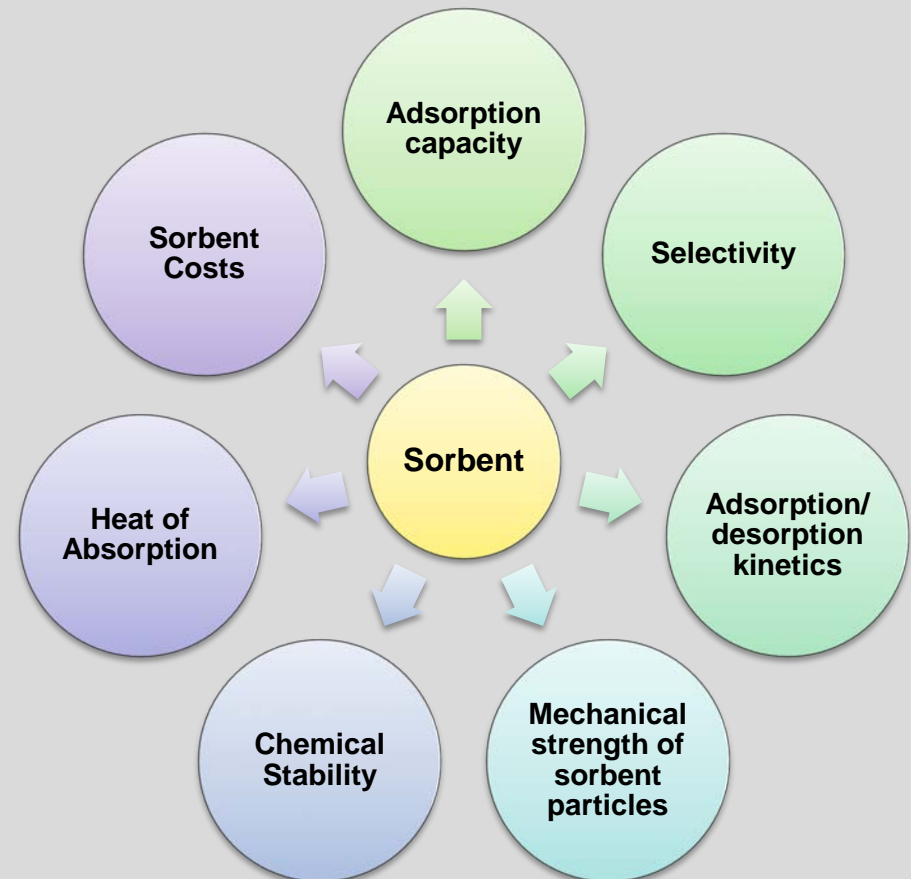
## BP2

- **Evaluate the impact of flue gas contaminants such as SO<sub>x</sub>, NO<sub>x</sub>, O<sub>2</sub>, and H<sub>2</sub>O on these advanced solids sorbents**
- **Conduct a high level techno-economic analysis.**

# Selection Criteria for CO<sub>2</sub> Solid Sorbents

Develop and design CO<sub>2</sub> capture solid sorbent that is chemically, thermally, and physically stable over multiple absorption/regeneration cycles and shows significant potential to meet the DOE program targets for CO<sub>2</sub> capture (>90% CO<sub>2</sub> capture rate with 95% CO<sub>2</sub> purity and <30% increase in cost of electricity).

- Fluidizable material
- High CO<sub>2</sub> loadings, high selectivity
  - $\geq 12$  wt% CO<sub>2</sub> capture
- No PEI leaching or degradation
  - Thermal & Oxidative stability
- Low heat of absorption
- Acceptable density
  - Density  $\geq 0.6$  to 1 g/cc
- Acceptable attrition resistance
  - Low makeup rate
- Economically practical
  - Low cost and easy scalability



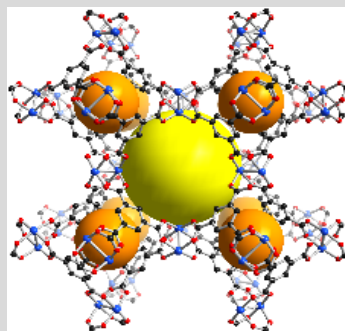
# Hybrid MOF-Based CO<sub>2</sub> Adsorbents

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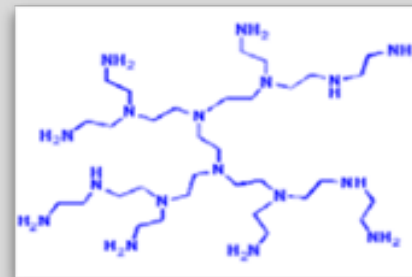
**Silica**

- Attrition resistance
- Fluidizable
- Low cost
- Acceptable density



**MOF (HKUST-1)**

- Exceptionally high surface areas
- Tunable pore sizes
- Commercially available linker

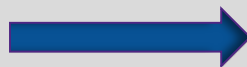


**PEI**

- High amine content
- High CO<sub>2</sub> affinity
- Relatively low cost materials



**Silica + MOF + PEI**



**Table 2. Reported CO<sub>2</sub> Capture Performance Results**

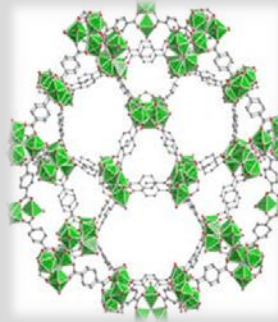
Sample Description	CO <sub>2</sub> Capacity (wt%)
MOF	21.4 <sup>a</sup>
MOF-amine	~20
Fluidizable silica (FS)	0
FS-PEI	4.8
MOF-silica	0
MOF-silica-amine	9.3

<sup>a</sup>Final report award # DE-FC26-07NT43092

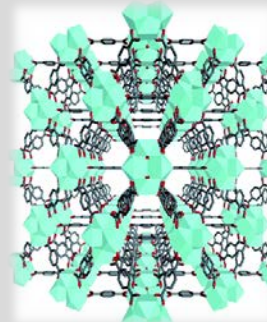
# MOFs Selected for Evaluation as Hybrid MOF-Based CO<sub>2</sub> Adsorbents

- Air and water stability
- Chemical Stability
- High thermal stability
- High selectivity for CO<sub>2</sub> over other components in flue gas (N<sub>2</sub> and O<sub>2</sub>)
- Commercially available linkers

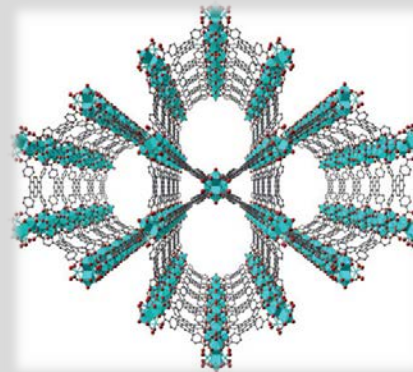
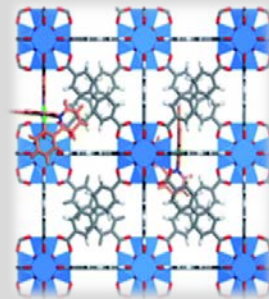
MIL101 (Cr)



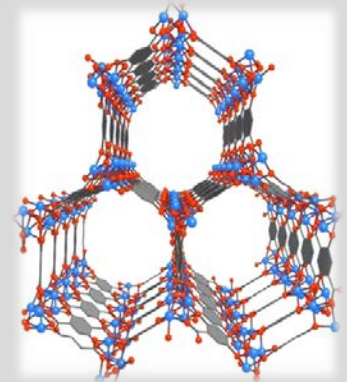
UIO-66 (Zr)



UIO-67 (Zr)



NU-1000 (Zr)

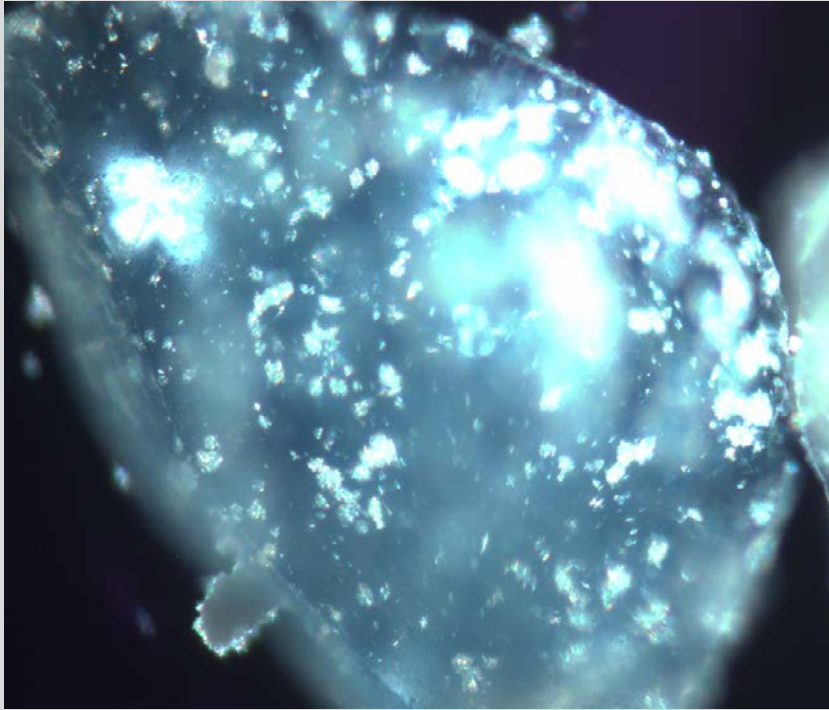


MOF-74 (Mg)

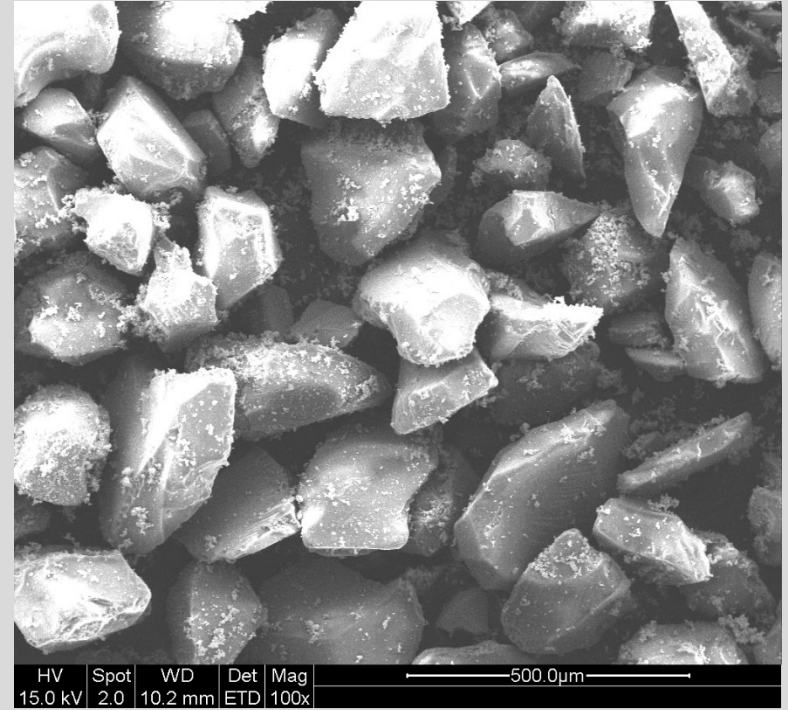
**Growing MOF inside the pores of Silica!**

# Solvothermal Synthesis of MOF-Silica Hybrid

The State-of-Art Solvothermal Synthesis of MOF-Silica Hybrid is non-selective!



Confocal microscope picture



SEM picture

Is the current solvothermal method the best approach for the MOF-Silica hybrid synthesis?

- Not utilizing the internal pores of the silica
- Poor interaction of MOF with Silica → Low yields
- Low attrition-resistance

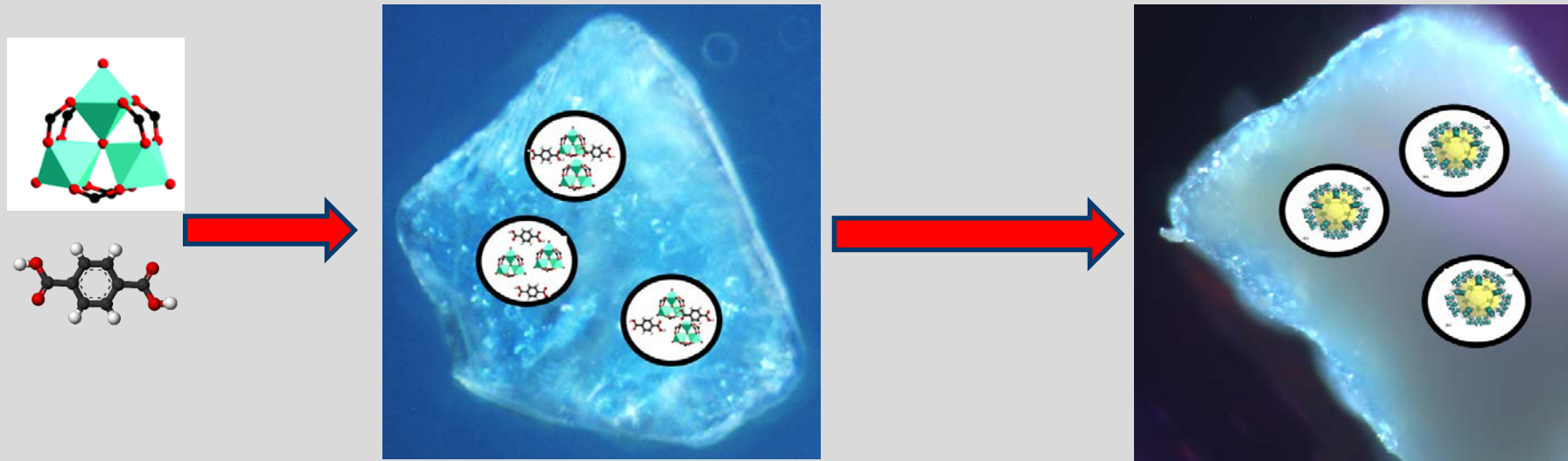


# A need for a New Approach!

- **Exhibit high MOF loading within the pores of silica ( $\text{SiO}_2$ )**
- **Excellent MOF dispersion and homogeneity**
- **Elevated surface area as *hybrid MOF-Silica***
- **Nanometric MOF particles**
- **Good Fluidizability**
- **Good handling**

# New Approach for MOF-Silica Hybrid Preparation

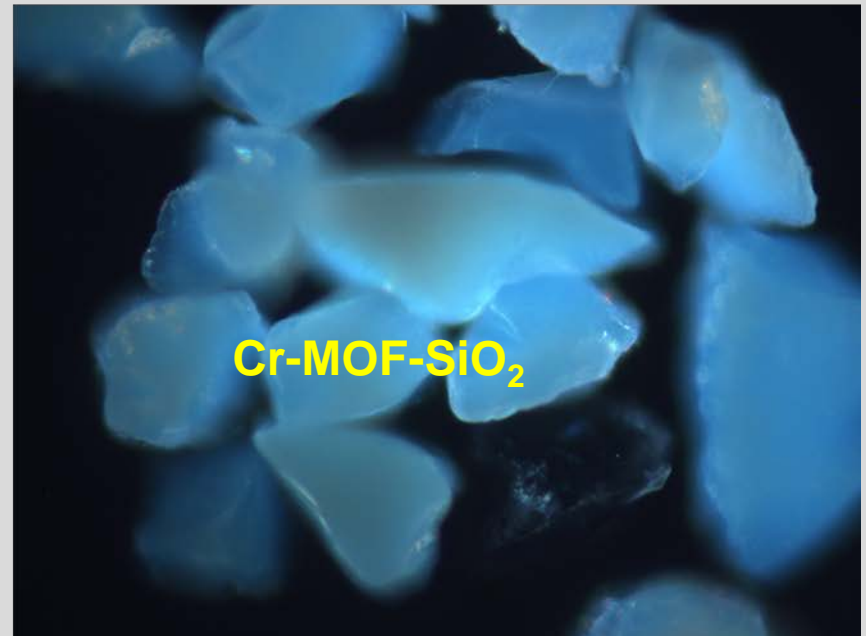
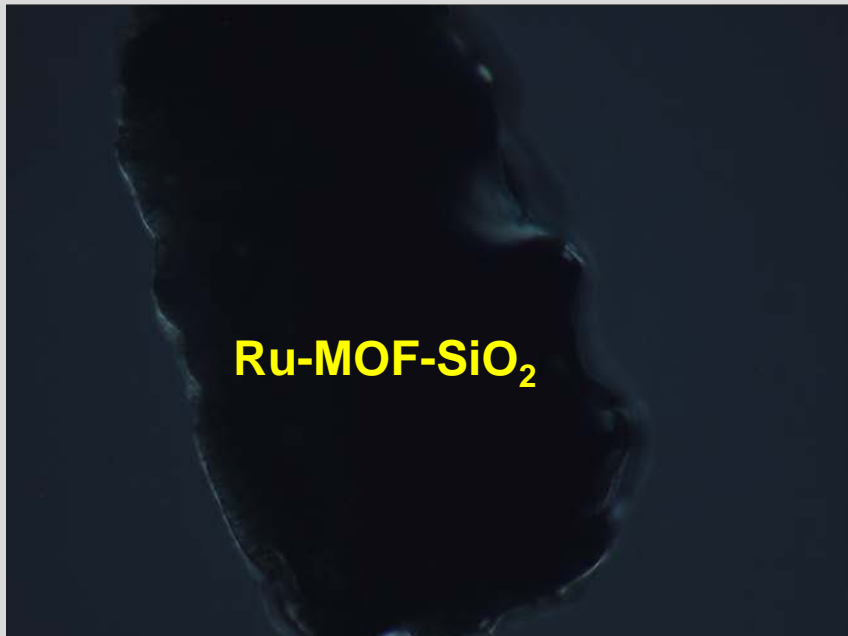
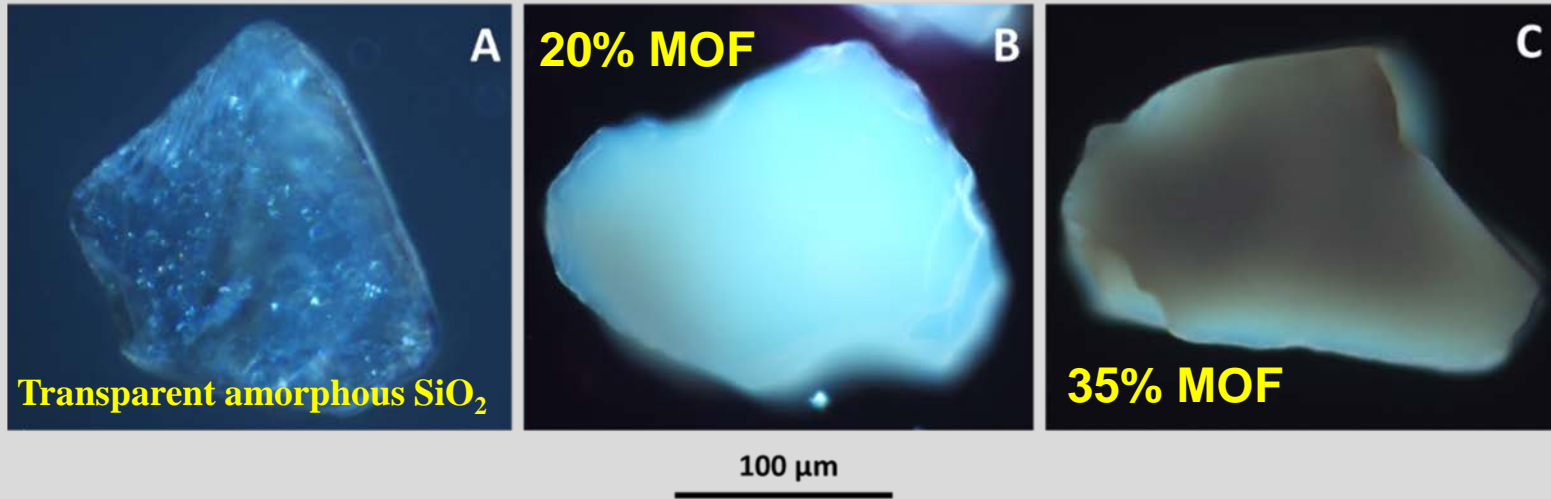
## Our new approach: Solid State Synthesis



New approach allowed the project to meet the first goal of the MOF-Silica hybrid Synthesis

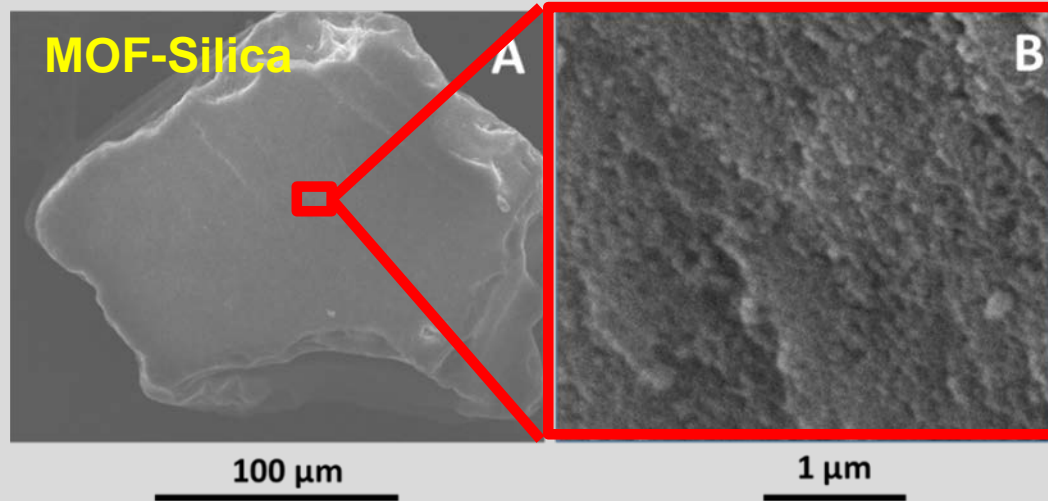
Full characterization using the most well known technics such as: Confocal Microscope, SEM, FIB-FESEM, TEM, FTIR, XRD, XRF, N<sub>2</sub> isotherms, TGA, Particle size distribution, Jet-Cut attrition index

# Confocal Microscope for the New MOF-Silica Hybrids

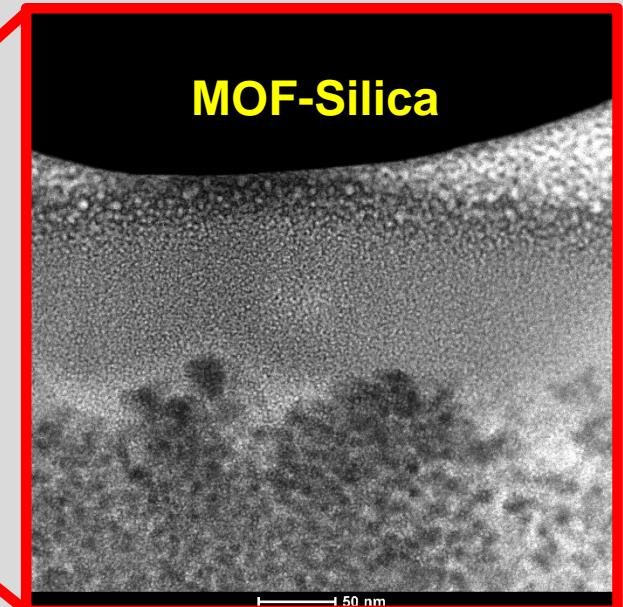
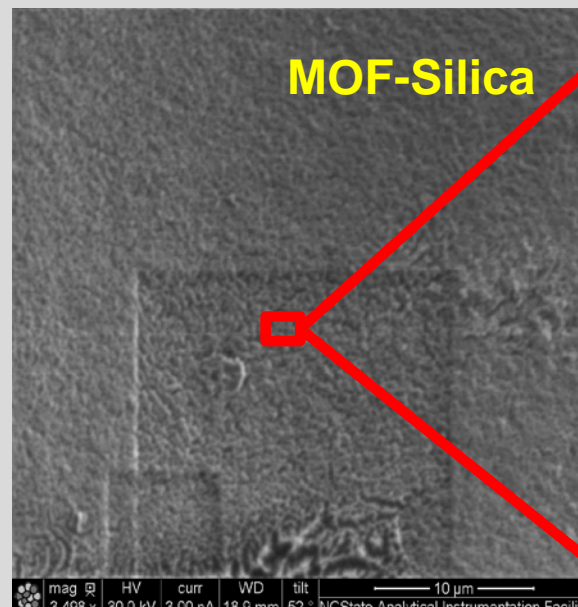
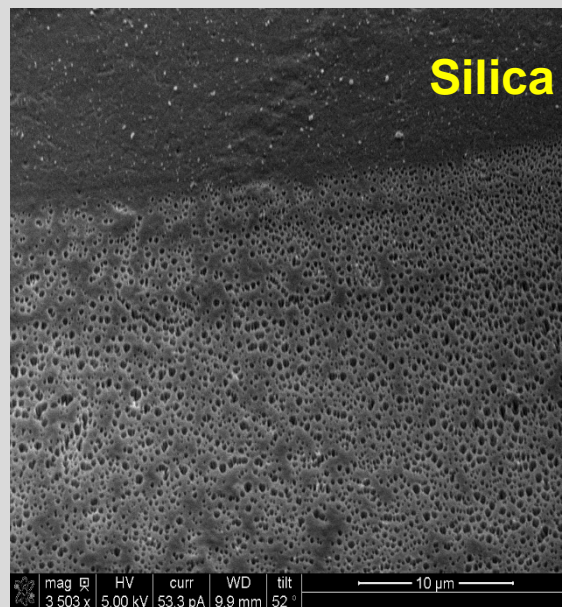


# Scanning Electron Microscope (SEM & FIB-FESEM)

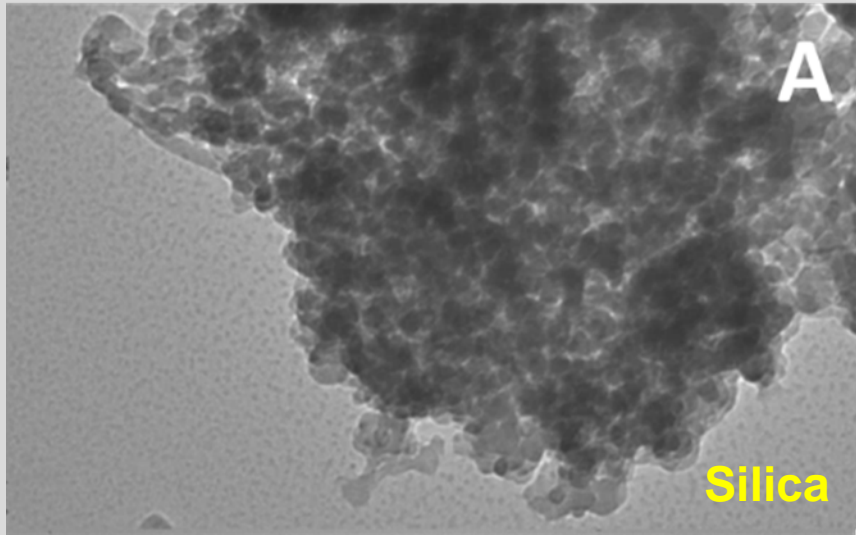
## SEM Pictures



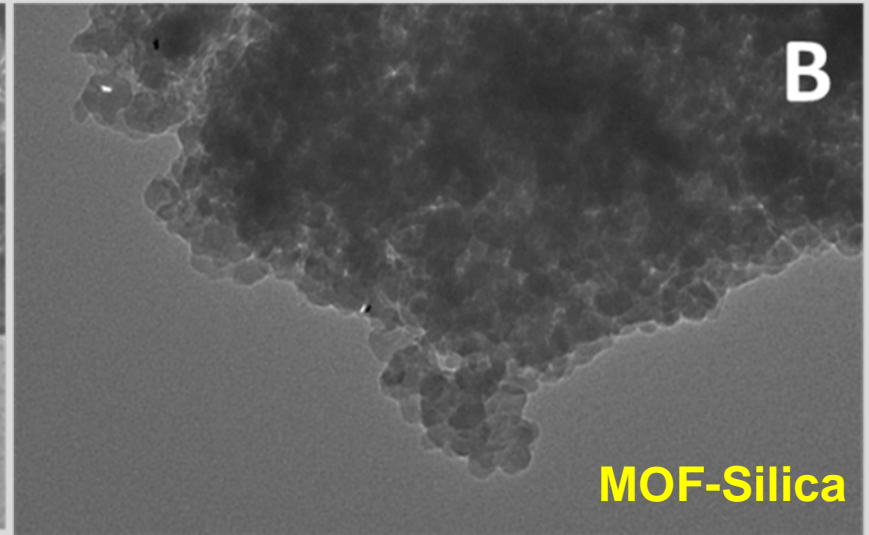
## FIB-FESEM Pictures



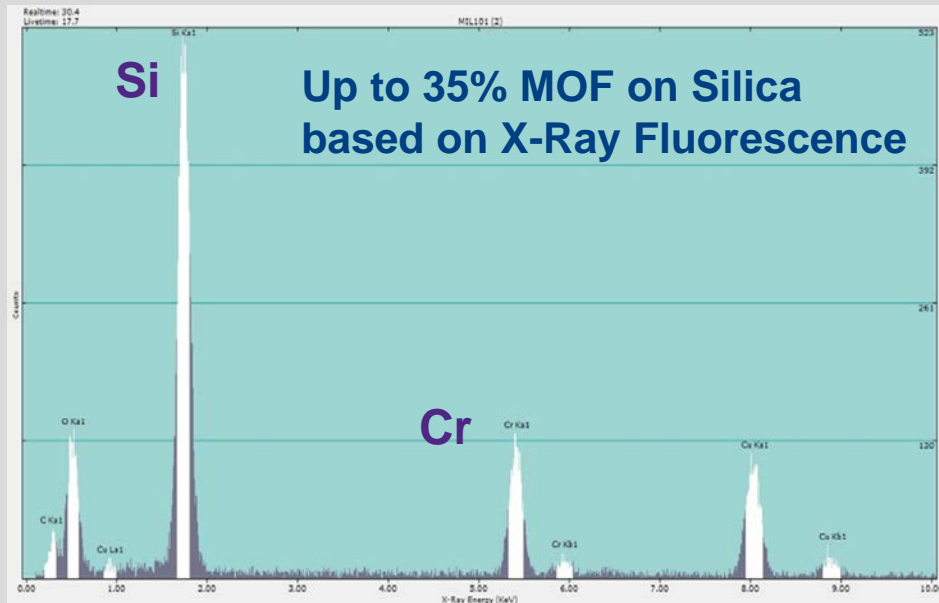
# Transmission Electron Microscopy (TEM) & X-Ray Fluorescence (XRF)



100 nm

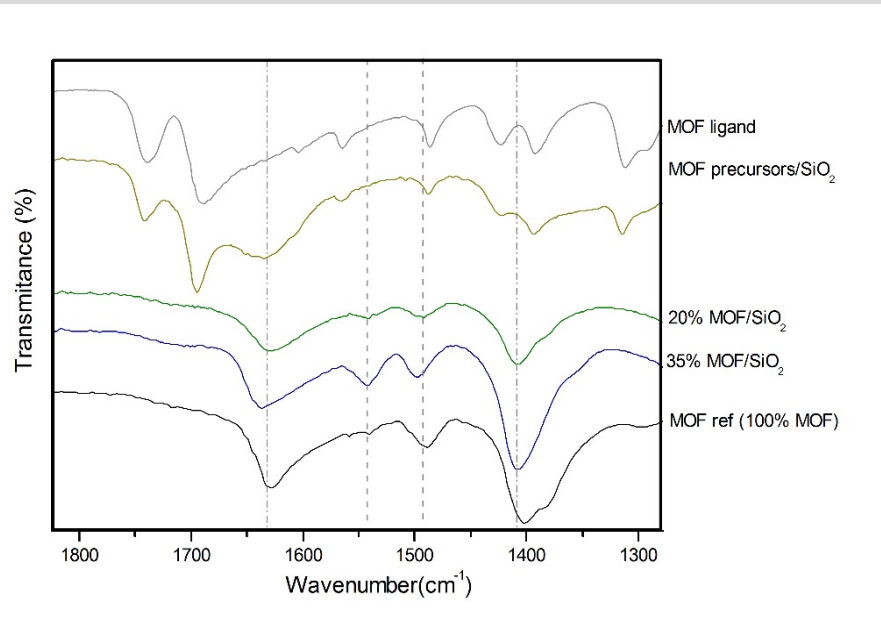


100 nm

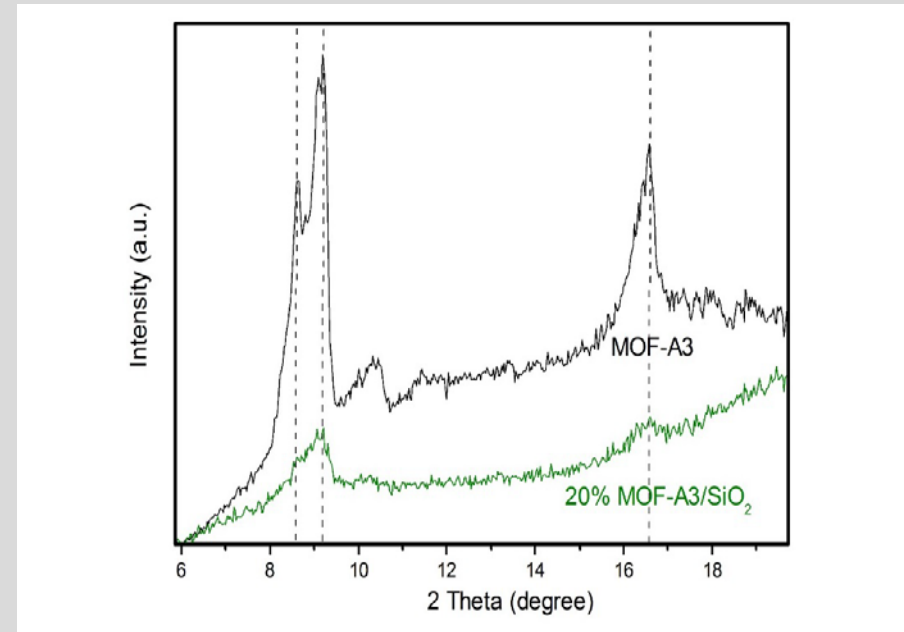


# Characterization of The New MOF-Silica Hybrid

## FTIR data

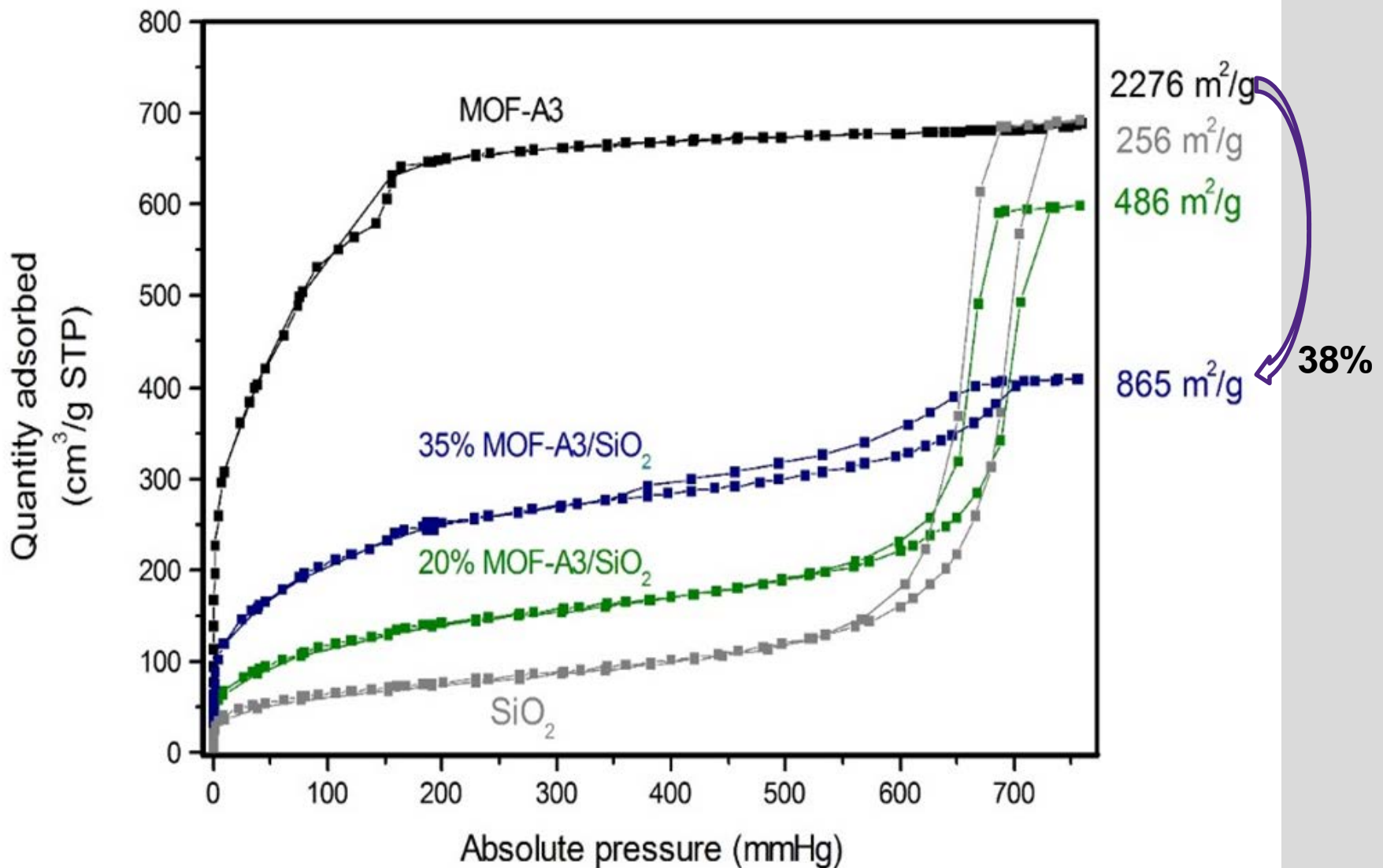


## X-Rays diffraction (XRD)



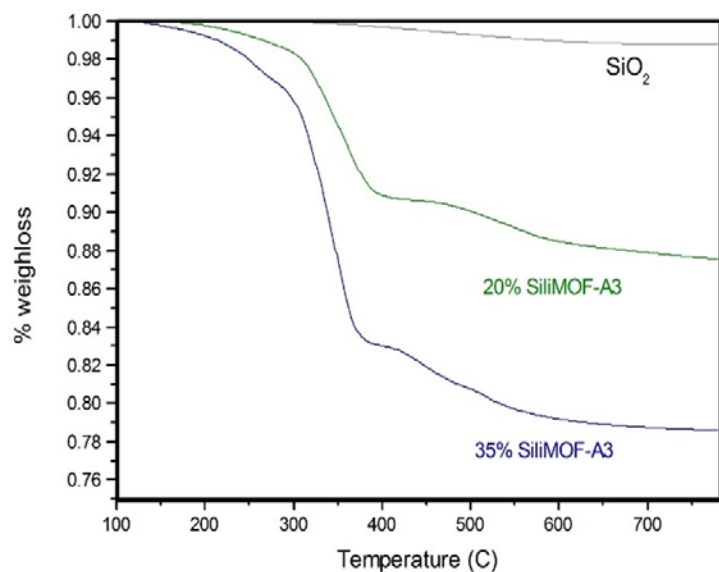
# Characterization of The New MOF-Silica Hybrid

## N<sub>2</sub> isotherms

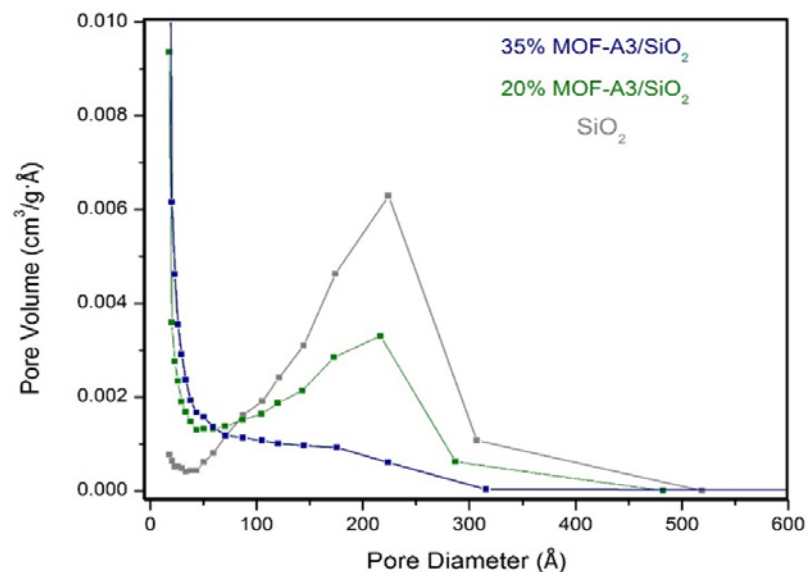


# Characterization of The New MOF-Silica Hybrid

## Thermogravimetric analysis



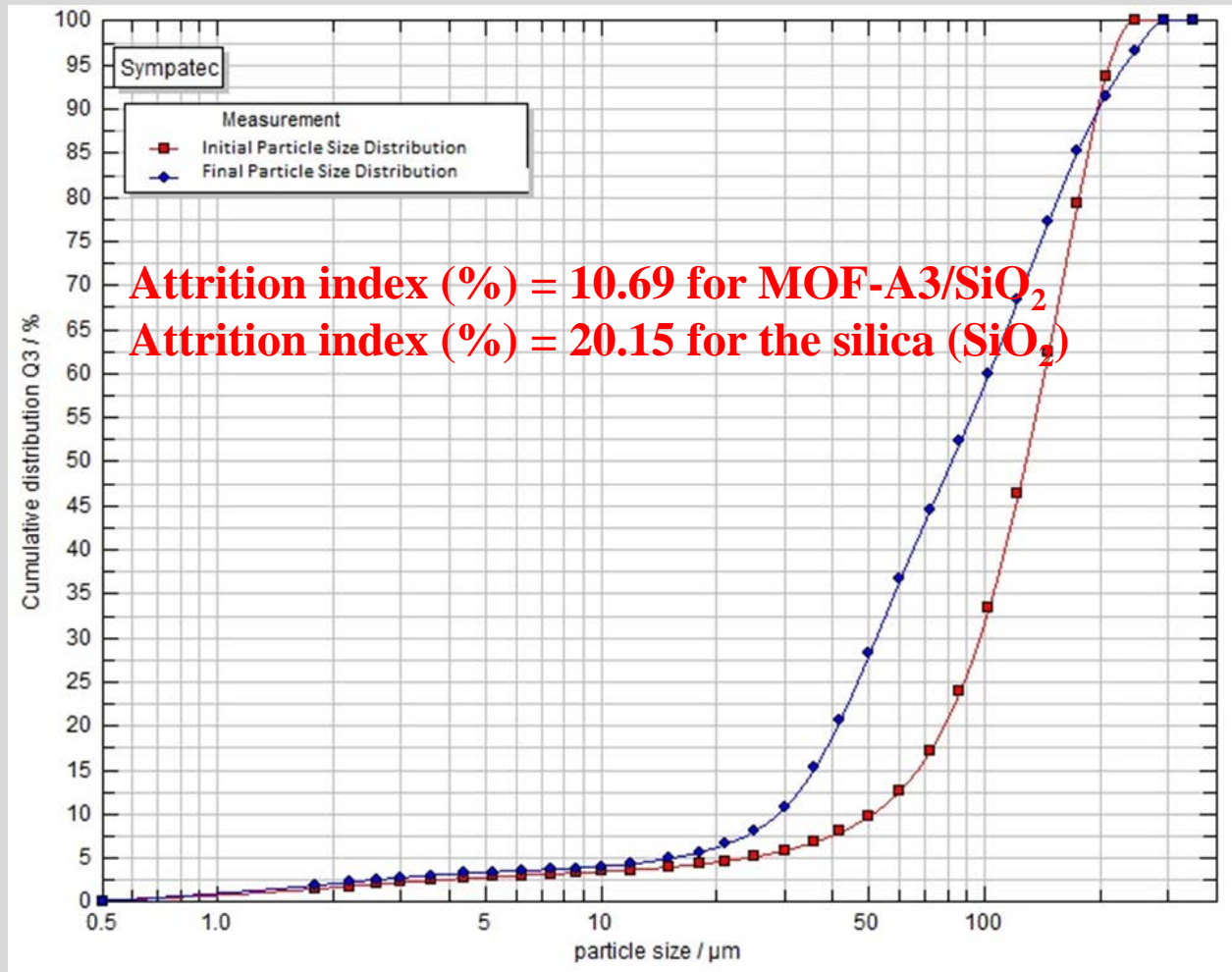
## Pore size distribution





# Jet Cup attrition index For MOF-Silica Hybrid

Comparison between initial particle size distribution and final particle size distribution for MOF-A3/SiO<sub>2</sub> using Jet Cup attrition index



Average Particle size Distribution is 159  $\mu\text{m}$  for the MOF-A3/SiO<sub>2</sub>

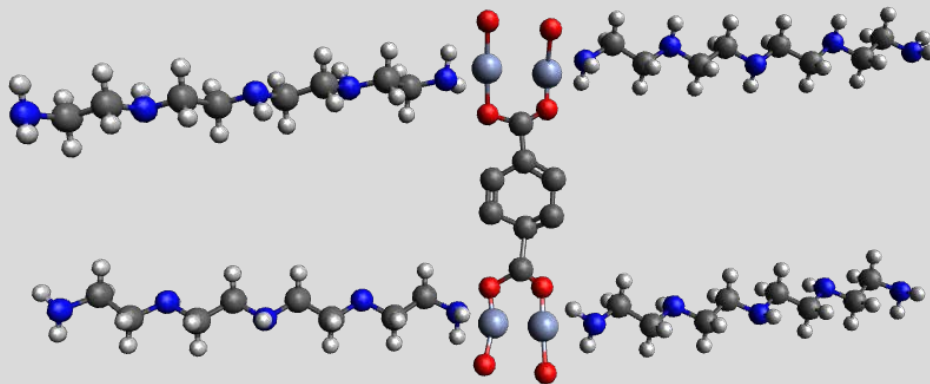
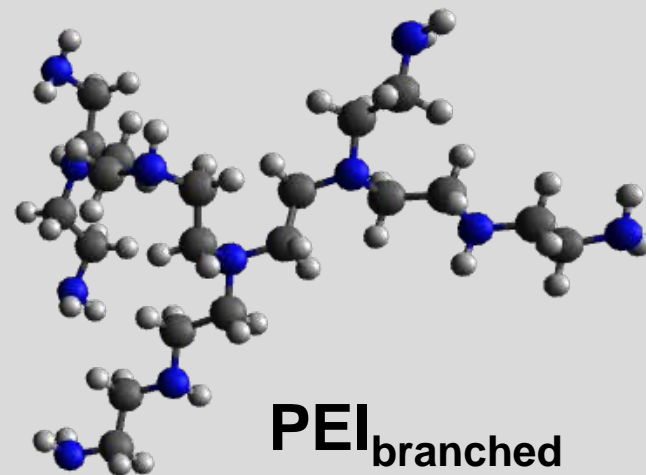
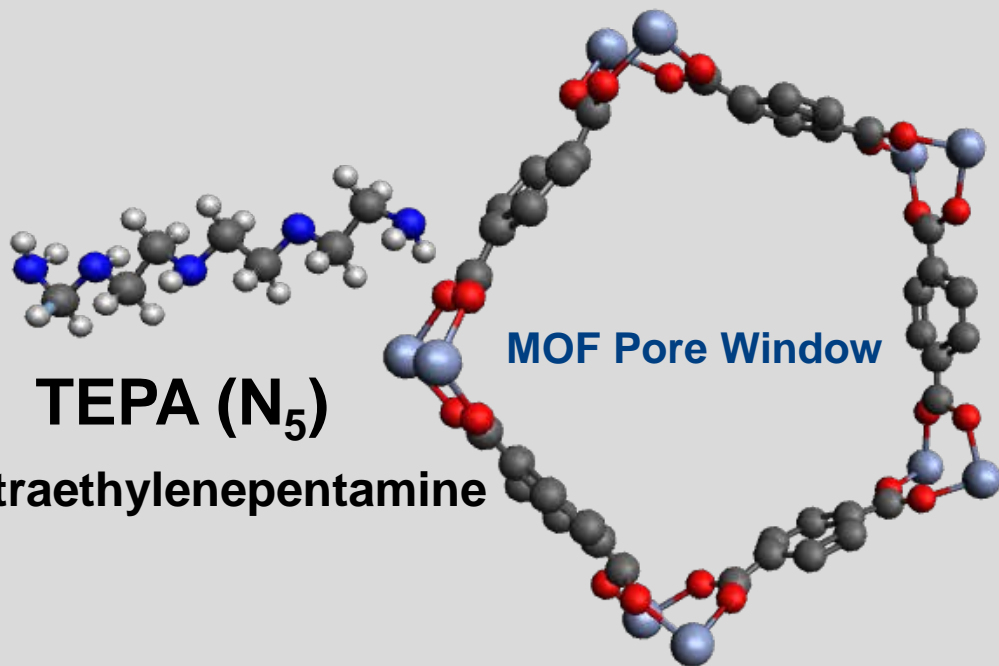
# Hybrid MOF-Silica Properties

- Exhibit high MOF loading (up to 35% so far)
- Excellent MOF dispersion and homogeneity
- Tunable hierarchical micro (MOF)-meso ( $\text{SiO}_2$ ) pore size distribution
- Elevated surface areas (up to  $900 \text{ m}^2/\text{g}$ )
- Nanometric MOF particles (below 30 nm, enhanced diffusion by shortening the diffusion channels)
- Enhanced attrition resistance
- Good fluidizability
- Good handling (100-500 microns).



**Hybrid MOF-Silica Sorbent (Amine addition)**

# Hybrid MOF-Based CO<sub>2</sub> Adsorbents

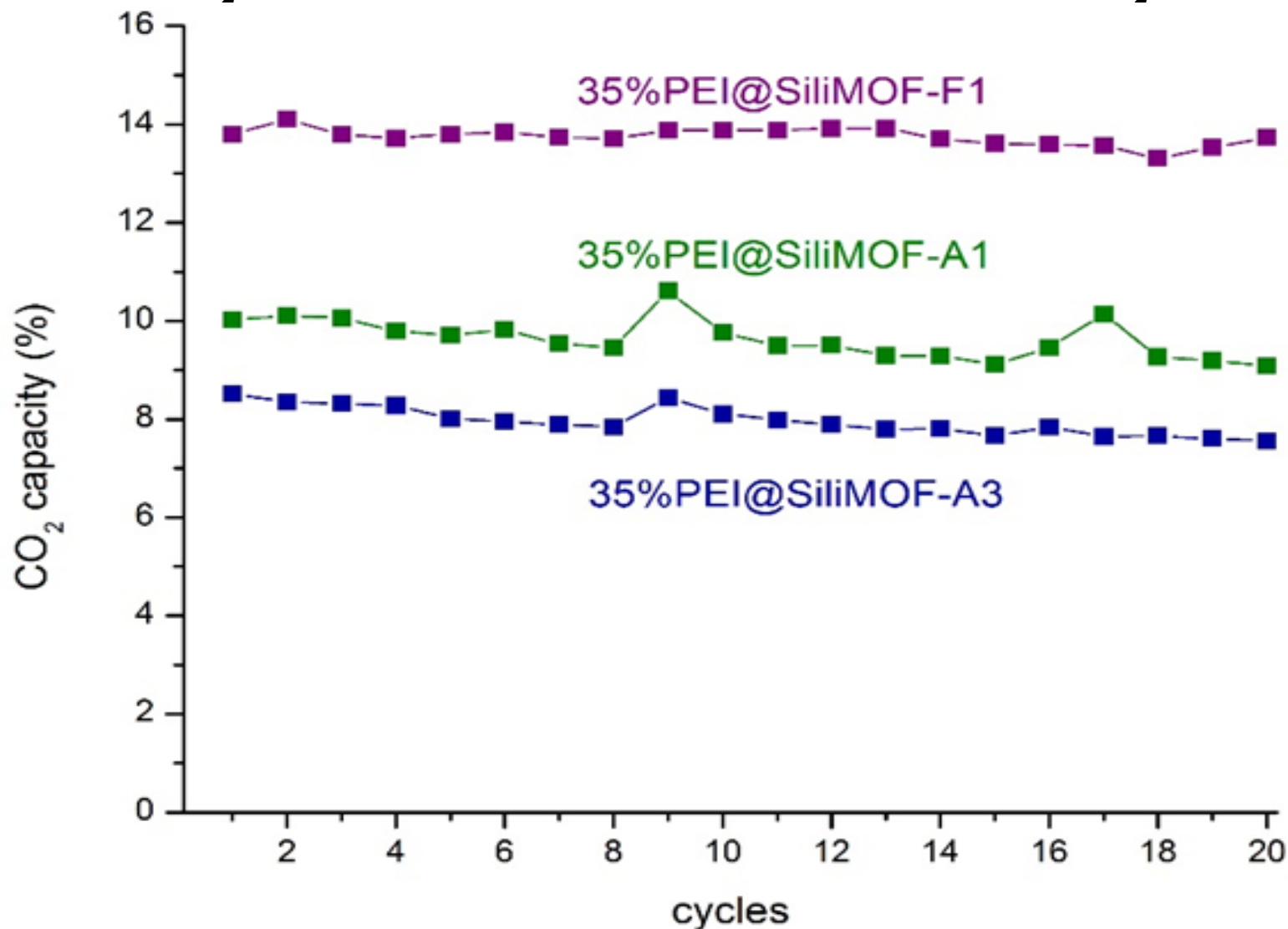


Different amines have been used taking advantage of the coordinatively unsaturated sites (CUS) and demonstrated that the amine stays inside the hybrid material after washing.

# Hybrid MOF-Based CO<sub>2</sub> Adsorbents Performance in RTI-PBR

**Simulated flue gas composition:**

**CO<sub>2</sub> = 15 vol%, O<sub>2</sub> = 4.5 vol%, and water = 5.65 vol% balanced with N<sub>2</sub>**



# Hybrid MOF-Based CO<sub>2</sub> Adsorbents

- We have prepared several MOFs based on open-literature synthetic methods
- We have grown MOF inside the pores of silica and collected a clear scientific data that supports MOF@Silica.
  - We developed a very elegant, novel and environmentally friendly way of growing MOF inside the pores of silica that could be extended to other MOFs.
- We have impregnated this hybrid materials (MOF@Silica) with different amines taking advantage of the coordinatively unsaturated sites (CUS) and demonstrated that the amine stays inside the hybrid material after washing.
- We have shown high CO<sub>2</sub> capacity ( $\geq 14$  wt.%) coupled with a good stability of this novel hybrid MOF-based CO<sub>2</sub> adsorbent
- We are in the process of optimizing the synthesis of these hybrid MOF-based CO<sub>2</sub> adsorbents and extending this finding to other hybrid MOF-based CO<sub>2</sub> adsorbents.

# Acknowledgements



**The U.S. DOE/National  
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**Steve Mascaro** (NETL  
Project Manager)



**State of North Carolina**



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